Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Francesco Punzo,^a*‡ David J. Watkin,^b Michela Iezzi Simone^c and George W. J. Fleet^c

^aDipartimento di Scienze Chimiche, Facoltà di Farmacia, Università di Catania, Viale A. Doria 6, 95125 Catania, Italy, ^bDepartment of Chemical Crystallography, Chemical Research Laboratory, University of Oxford, Mansfield Road, Oxford OX1 3TA, England, and ^cDepartment of Organic Chemistry, Chemical Research Laboratory, University of Oxford, Mansfield Road, Oxford OX1 3TA, England

Visiting Scientist at the Department of Chemical Crystallography, Chemical Research Laboratory, University of Oxford, Mansfield Road, Oxford OX1 3TA, England.

Correspondence e-mail: francesco.punzo@chemistry.oxford.ac.uk

Key indicators

Single-crystal X-ray study T = 120 K Mean σ (C–C) = 0.003 Å R factor = 0.034 wR factor = 0.077 Data-to-parameter ratio = 9.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The title compound, $C_{15}H_{20}N_2O_5$, is the first example of a branched tetrahydrofuran sugar amino acid dipeptide isostere incorporated into a peptidomimetic. The crystal structure contains intermolecular $N-H\cdots O$ and $O-H\cdots O$ hydrogen

(3R,4R,5S)-5-(Acetamidomethyl)-N-benzyl-3,4-

dihydroxytetrahydrofuran-3-carboxamide

Received 10 December 2004 Accepted 14 January 2005 Online 29 January 2005

Comment

bonds.

 δ -Tetrahydrofuran (THF) sugar amino acids (SAA) have been extensively investigated as dipeptide isosteres (Baron et al., 2004; Grotenberg et al., 2004; Raunkjr et al., 2004). Introduction of δ -THF SAA building blocks has been shown to induce secondary structural features such as β -turn-like structures (Chakraborty et al., 2004; Smith et al., 2003; Hungerford et al., 2000) and helices (Claridge et al., 1999; Osterkamp et al., 2000) in small peptidomimetics. All the previously reported δ -THF SAA scaffolds have linear carbon chains, as in (1), which has been incorporated into peptidomimetics such as (2). The synthesis of branched sugar lactones (Hotchkiss et al., 2004) has allowed ready access to a new class of δ -THF SAA building blocks, such as (3), which contain a branched carbon chain. The monomer (3) was prepared as an oil from L-lyxonolactone in a sequence in which the branched carbon chain was introduced by the Ho (1978, 1985a,b) crossed aldol procedure, and the δ -THF ring was subsequently formed by an intramolecular alkylation. The branched scaffold (3) was transformed into the crystalline branched peptidomimetic (4).



The structure of (4) has been determined in order to remove any ambiguity in the stereochemical outcomes of either the aldol or the ring closure reactions. Additionally, the crystal structure of (4) may give some indication of the



The molecular structure of (4), with displacement ellipsoids drawn at the 50% probability level.

© 2005 International Union of Crystallography

Printed in Great Britain - all rights reserved

organic papers

secondary structural motif likely to be induced by the incorporation of the monomer (3) into peptidomimetics. The molecular structure of (4) is shown in Fig. 1. As usually expected for sugar derivatives, there are intermolecular hydrogen bonds (Table 2 and Fig. 2).

Experimental

Compound (4) was dissolved in acetone in a small glass cylinder and then crystallized as the solvent evaporated slowly to give colourless needle-like crystals.

Crystal data

 $C_{15}H_{20}N_2O_5$ $M_r = 308.33$ Orthorhombic, $P2_12_12_1$ a = 15.3802 (6) Å b = 5.4473 (2) Å c = 18.0635 (8) Å V = 1513.37 (10) Å³ Z = 4 $D_x = 1.353$ Mg m⁻³

Data collection

Nonius KappaCCD diffractometer ω scans Absorption correction: multi-scan (*DENZO/SCALEPACK*; Otwinowski & Minor, 1997) $T_{min} = 0.820, T_{max} = 0.998$ 5747 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.034$ $wR(F^2) = 0.077$ S = 0.931973 reflections 200 parameters H-atom parameters constrained Cell parameters from 3224 reflections $\theta = 5-27^{\circ}$ $\mu = 0.10 \text{ mm}^{-1}$ T = 120 KNeedle, colourless $0.40 \times 0.04 \times 0.02 \text{ mm}$

Mo $K\alpha$ radiation

1973 independent reflections 1594 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.048$ $\theta_{\text{max}} = 27.4^{\circ}$ $h = -19 \rightarrow 19$ $k = -7 \rightarrow 5$ $l = -23 \rightarrow 23$

 $w = 1/[\sigma^{2}(F^{2}) + 0.02 + 0.04P],$ where $P = [\max(F_{o}^{2}, 0) + 2F_{c}^{2}]/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.27 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.23 \text{ e } \text{\AA}^{-3}$ Extinction correction: Larson (1970) Extinction coefficient: 16 (5)

Table 1

Selected bond lengths (Å).

C1-C2	1.551 (2)	C6-N7	1.453 (2)
C1-C5	1.525 (3)	N7-C8	1.334 (2)
C1-C12	1.531 (3)	C8-O9	1.236 (2)
C1-O22	1.421 (2)	C8-C10	1.501 (3)
C2-C3	1.518 (3)	C12-N13	1.333 (2)
C2-O11	1.421 (2)	C12-O21	1.235 (2)
C3-O4	1.438 (2)	N13-C14	1.454 (2)
C3-C6	1.522 (3)	C14-C15	1.515 (3)
O4-C5	1.436 (2)		

Ta	bl	е	2
----	----	---	---

H	ſyd	lrogen-	bond	geometry	(A	., °))
---	-----	---------	------	----------	----	--------------	---

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
022-H6···O9 ⁱ	0.95	1.75	2.649 (2)	158
$N7-H12 \cdot \cdot \cdot O21^{ii}$	1.00	1.95	2.953 (2)	177
$O11-H1\cdots O11^{iii}$	0.95	1.95	2.886 (2)	166
Symmetry codes:	(i) $x - \frac{1}{2}, -y$	$+\frac{3}{2}, -z+2;$	(ii) $-x + 2, y + 3$	$\frac{1}{2}, -z + \frac{3}{2};$ (iii)

Symmetry codes: (i) $x - \frac{1}{2}, -y + \frac{3}{2}, -z + 2;$ (ii) $-x + 2, y + \frac{1}{2}, -z + \frac{3}{2};$ (iii) $-x + 2, y - \frac{1}{2}, -z + \frac{3}{2};$





A packing diagram of (4), viewed down the *b* axis. Hydrogen bonds are indicated by dashed lines.

In the absence of significant anomalous scattering, Friedel pairs were merged. The absolute configuration of (4) was assigned since the starting material was L-lyxonolactone with known absolute configuration and two of the chiral centres are retained (see scheme). H atoms were located in difference density maps. Those attached to C atoms were repositioned geometrically. The H atoms were initially refined with soft restraints on the bond lengths and angles to regularize their geometry (C-H = 0.98–1.01 Å, O-H = 0.95 Å and N-H = 0.95–1.00 Å), after which they were refined as riding, with $U_{\rm iso}({\rm H}) = 1.2U_{\rm eq}({\rm C})$, and $U_{\rm iso}({\rm H}) = 0.05 Å^2$ for those bonded to N and O atoms.

Data collection: *COLLECT* (Nonius, 2001); cell refinement: *DENZO/SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO/SCALEPACK*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *CRYSTALS*.

Financial support (to MIS) provided through the European Community's Human Potential Programme under contract HPRN-CT-2002-00173 is gratefully acknowledged.

References

- Altomare, A., Cascarano, G., Giacovazzo, G., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). J. Appl. Cryst. 27, 435–435.
- Baron, R., Bakowies, D. & van Gunsteren, W. F. (2004). Angew. Chem. Int. Ed. 43, 4055–4059.
- Betteridge, P. W., Carruthers, J. R., Cooper, R. I., Prout, K. & Watkin, D. J. (2003). J. Appl. Cryst. 36, 1487.
- Chakraborty, T. K., Srinivasi, P., Tapadar, S. & Mohan, B. K. (2004). J. Chem. Sci. 116, 187–207.

Claridge, T. D. W., Long, D. D., Hungerford, N. L., Aplin, R. T., Smith, M. D., Marquess, D. G. & Fleet, G. W. J. (1999). *Tetrahedron Lett.* **40**, 2199–2202.

- Grotenberg, G. M., Timmerj, M. S. M., Llamas-Saiz, A. L., Verdoes, M., van der Marel, G. A., van Raaij, M. J., Overkleeft, H. S. & Overhand, M. (2004). J. Am. Chem. Soc. 126, 3444–3446.
- Ho, P. T. (1978). Tetrahedron Lett. 19, 1623-1626.
- Ho, P. T. (1985a). Can. J. Chem. 57, 381-381.
- Ho, P. T. (1985b). Can. J. Chem. 63, 2221-2224.
- Hotchkiss, D., Soengas, R., Simone, M. I., van Ameijde, J., Hunter, S., Cowley, A. R. & Fleet, G. W. J. (2004). *Tetrahedron Lett.* **45**, 9461–9464.

Hungerford, N. L., Claridge, T. D. W., Watterson, M. P., Aplin, R. T., Moreno, A. & Fleet, G. W. J. (2000). J. Chem. Soc. Perkin Trans. 1, 21, 3666–3679.

Larson, A. C. (1970). Crystallographic Computing, edited by F. R. Ahmed, pp. 291–294. Copenhagen: Munksgaard.

- Nonius (2001). COLLECT. Nonius BV, Delft, The Netherlands. Osterkamp, F., Ziemer, B., Koert, U., Wiesner, M., Raddatz, P. & Goodman, S. L. (2000). Chem. Eur. J. 6, 666-683.
- Otwinowski, Z. & Minor, W. (1997). Methods in Enzymology, Vol. 276, Macromolecular Crystallography, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307-326. New York: Academic Press.
- Raunkjr, M., El Oualid, F., van der Marel, G. A., Overkleeft, H. S. & Overhand, M. (2004). Org. Lett. 6, 3167-3170.
- Smith, M. D., Claridge, T. D. W., Sansom, M. P. & Fleet, G. W. J. (2003). Org. Biomol. Chem. 1, 3647-3655.
- Watkin, D. J., Prout, C. K. & Pearce, L. J. (1996). CAMERON. Chemical Crystallography Laboratory, Oxford, England.